**Titration Lab**

**Problems:**

1. **How can you neutralize 2 different acids?**
2. **How can you use the volumes of the base required to neutralize the 2 acids to determine which acid was more concentrated (higher Molarity)?**

**Material: 2 acids 10 ml, base, graduated cylinder, beakers and pipettes.**

**Data:**

|  |  |  |  |
| --- | --- | --- | --- |
| **Acid** | **pH** | **pH after Base is added** | **Volume of base used (ml)** |
| **1.)** |  |  |  |
| **2.)** |  |  |  |

**Conclusion: From notes**