Solubility table labII

Problems:

1. How can you determine if Lead Hydroxide is soluble or not?
2. How can you make Lead Hydroxide?
3. How can you write the chemical equation for the reaction that created Lead Hydroxide?

Material: NaOH, Pb(NO3)2, periodic table, test tubes.

Data:

|  |  |  |
| --- | --- | --- |
|  | Will you see a reaction? | Did you see a reaction? |
| NaOH (5ml) + Pb(NO3)2 (5ml) |  |  |

Ion exchange:

Conclusion: From Notes.