Surface Tension and

Polar Molecules.

Surface Tension: is the attraction between molecules. The stronger the surface tension the less likely the molecules will spread out.

Polar Molecules: are molecules that have a positive and a negative end due to the shape and oxidation charges of the atoms that make the molecule.

Example: water

* -

Positive pole

+ +

**\_ \_**

Negative pole + +

**The stronger the attraction between oppositely charged poles the more polar the molecule and the stronger the surface tension.**