**Neutralization Lab**

**Problems:**

1. **How can you use Universal Indicator to determine the pH of an Acid and a   
   Base?**
2. **What will happen to the pH of the acid when you mix it with the mase?**

**Material: H2SO4 (10ml), NaOH (10 ml), Beakers, UI.**

**Data:**

|  |  |  |  |
| --- | --- | --- | --- |
|  | **H2SO4** | **NaOH** | **H2SO4 + NaOH** |
| **pH** |  |  |  |
| **Acid or Base?** |  |  |  |

**Conclusion: ?’s from notes.**