Molarity Problems

II

1. **If it took 15.98 ml of .5 M Mg(OH)2 to neutralize 9.45 ml of an unknown Hydroflouric acid, what is the molarity of the acid?**
2. **If it took 55.5 ml of an unknown Nitric acid to neutralize 22.5 ml of a 2.5 M Lead (III) hydroxide, what is the molarity of the acid?**